Fermi-Löwdin orbitals for SIC Recovering chemical concepts / improved correlation energies

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Self interaction-correction with Fermi-Löwdin orbitals

The self-interaction error

$$E_H = \frac{1}{8\pi\epsilon} \int d^3\mathbf{r} \int d^3\mathbf{r}' \frac{\rho(\mathbf{r})\rho(\mathbf{r}')}{|\mathbf{r} - \mathbf{r}'|}$$

- E_H should be zero for single electron
 → but it is not !
- Perdew, Zunger (1981):

$$E_{XC}^{PZ-SIC} = -\sum_{i,\sigma} \{ U\left[\rho_{i,\sigma}\right] + E_{XC}^{approx}\left[\rho_{i,\sigma},0\right] \}$$
$$\{ H_{0,\sigma} + V_{i,\sigma}^{SIC} \} \left|\psi_{i,\sigma}\right\rangle = \sum_{j} \lambda_{ij}^{\sigma} \left|\psi_{j,\sigma}\right\rangle$$

 Orbital dependent functional leads to excessive computational demands

Perdew, J. P., Zunger, A., Phys. Rev. B, 23 5048 (1981).

Fermi-Löwdin orbital SIC

$$F_{i\sigma}(\mathbf{r}) = \frac{\sum_{\alpha} \psi_{\alpha\sigma}^*(\mathbf{a}_{i\sigma})\psi_{\alpha\sigma}(\mathbf{r})}{\sqrt{\sum_{\alpha} |\psi_{\alpha\sigma}(\mathbf{a}_{i\sigma})|^2}}$$

- 1. Transform KS-orbitals Ψ to Fermi-orbitals $F_{\rm i}$
- 2. Use Löwdin-orthogonalization $F_i \rightarrow \varphi_i$

$$\sum_{j} S_{ij} T_{\alpha j} = Q_{\alpha} T_{\alpha i} \rightarrow |\phi_k\rangle = \sum_{\alpha j} \frac{1}{\sqrt{Q}} T_{\alpha k} T_{\alpha j} |F_j\rangle$$

 Minimization of total energy incl. E_{SIC} (finding 3N reference positions a_i)

Pederson, M. R. et al., J. Chem. Phys. 140, 121103 (2014). Pederson, M. R., *J. Chem. Phys.*, *142*, 064112 (2015).

Density Functional Theory + FLO-SIC

-0.46

-0.48

-0.50

-0.56

-0.58

-0.60

2

1

3

5

r[H – H] (bohr)

6

7

8

$$F_{i\sigma}(\mathbf{r}) = \frac{\sum_{\alpha} \psi_{\alpha\sigma}^* (\mathbf{a}_{i\sigma}) \psi_{\alpha\sigma}(\mathbf{r})}{\sqrt{\sum_{\alpha} |\psi_{\alpha\sigma}(\mathbf{a}_{i\sigma})|^2}}$$

Gradients of FLO

SO42-



N (electrons)

Application to molecules

Improved ionization potentials from ϵ_{HOMO}

Correct level ordering



T. Hahn et al., J. Chem. Phys. 143, 224104 (2015).

Chemical insights

Fermi-Löwdin orbitals – a simple example



- Fermi orbital descriptors can be seen as electronic positions
- Descriptors of 1s core close to nuclear positions
- Single bond: rotational symmetric, double and triple bond: 'banana'-like shapes
- SO4²⁻: four equivalent, single bonded oxygen (Lewis picture)

Kohn-Sham- and Fermi-Löwdin orbitals of Benzen



- Fermi orbital descriptors can be seen as electronic positions
- No physical meaning of KS orbitals
- Shape and position of Fermi-orbitals tell bonding details
- 4n+2 rule

$$E_{\sigma}^{\mathrm{X}} = -\frac{1}{2} \int \mathrm{d}^{3}\mathbf{r} \ \rho_{\sigma}\left(\mathbf{r}\right) \int \mathrm{d}^{3}\mathbf{a} \ \frac{\left|\mathrm{F}_{\mathrm{i}\sigma}\left(\mathbf{r}\right)\right|^{2}}{\left|\mathbf{r}-\mathbf{a}\right|_{\mathrm{S}}}$$

(Anti-) aromatic compounds



Bond analysis in Boron clusters



- Yields natural description of bonds in terms of m center- n electron bonds
- Correctly covers multicenter bonds
- No extra parameters or analysis steps needed, its just the result of a FLO-SIC DFT calculation

■ Aromatic? → Yes (4n+2) !

What else ?

Fermi-orbitals – What else can we do with them?

Fermi orbital descriptors of atoms:



- FOD positions with minimum energy are "connected"
- Position of one descriptor defines position of the others
- → correlated, have nonlocal properties

Can we use this ?

Correlation correction with Fermi-Löwdin orbitals

1. Fermi-Löwdin orbitals yield single electron densities

$$n_k(\mathbf{r}) = |\phi_k(\mathbf{r})|^2$$
 $n_T(\mathbf{r}) = \sum_k n_k(\mathbf{r})$

- 2. Use LDA^{*} correlation subroutine to calculate E_C for the single electron densities
- 3. Subtract these energies from E_c of the total density

$$E_c^{\text{corr.}} = E_c^{\text{LDA}} \left[n_T(\mathbf{r}) \right] - \sum_k E_c^{\text{LDA}} \left[n_k(\mathbf{r}) \right]$$

Single electron system:

$$|\phi(\mathbf{r})|^2 = |\psi(\mathbf{r})|^2$$

 $n_T(\mathbf{r}) = n_k(\mathbf{r})$

$$E_c^{\text{corr.}} = 0$$

! exact result !

* ... or any other functional

Correlation correction with Fermi-Löwdin orbitals

Atoms and small molecules (energies in Ha)

	exact*	LDA	Error (%)	FLO-Corr.	Error (%)
He	-0.042	-0.111	164	-0.053	26
В	-0.095	-0.224	135	-0.107	12
Ν	-0.186	-0.426	130	-0.203	10
Ne	-0.394	-0.739	87	-0.458	16
Na	-0.399	-0.801	100	-0.440	8
Mg	-0.442	-0.887	100	-0.504	10
Ar	-0.731	-1.423	94	-0.828	14
N ₂	-0.549	-0.940	71	-0.613	11
LiF	-0.441	-0.827	87	-0.501	13

 Reduced MAE: 107 % → 13 %

 Computational cheap

 Applicable to any semilocal functional

^{*} D. P. ONeil, P.M.W. Gill, Mol. Phys., 103, 763 (2005)

Summary

Fermi-orbital based SIC

- Correct ~ 1/r behavior of potential
- has correct derivative discontinuity
- Unlike Hubbard-U or range separated hybrids:
 a_i are fully determined by variational principle

Chemical insight

- Yields information about bonding in terms of multi-center multi-electron bonds
- Single-, double-, triple-bond ? Just count descriptors!
- Integral part of the FLO-SIC DFT calculation

Improved correlation energies

- Single electron densities can be used to correct correlation energies
- Computational inexpensive





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DFG HA 5070/3

Charging of TCNQ Fermi-orbital picture

(b)

4

$$F_{i\sigma}(\mathbf{r}) = \frac{\sum_{\alpha} \psi_{\alpha\sigma}^* (\mathbf{a}_{i\sigma}) \psi_{\alpha\sigma}(\mathbf{r})}{\sqrt{\sum_{\alpha} |\psi_{\alpha\sigma}(\mathbf{a}_{i\sigma})|^2}}$$





	LDA	FLO- SIC	Exp*
Ν	-7.3	-10.6	-9.7
N-1	-1.8	-5.3	-3.8
N-2	+2.1	-1.0	

$\epsilon_{\rm HOMO}$ vs. –IP of TCNQ

^{*} W. Kaim and M. Moscherosch, Coord. Chem. Rev. 129, 157 (1994), Guo-Zu Z., L. Wang., J. Chem. Phys. **143**, 221102 (2015). 17 First principles description of charge transfer compounds ?

Ground state charge transfer compound

Donor Acceptor Δ δδ+ charge transfer LUMO LUMO HOMO HOMO

> ${f E}_{HOMO}(D)\cong {f E}_{LUMO}(A)$ ${f E}_{Fermi}(D)\gg {f E}_{Fermi}(A)$

The problem(s) in standard DFT



- No bond breaking
- Li has positive charge at large R
- HOMO (IP) / LUMO (EA) unreliable

First principles description of charge transfer compounds ?

Ground state charge transfer compounds



 $E_{HOMO}(D) \cong E_{LUMO}(A)$ $E_{Fermi}(D) \gg E_{Fermi}(A)$

The problem in standard DFT



- Qualitative correct behavior for FLO-SIC
- Bond breaking
- Neutral atom in infinite separations



Initial guess a_i

Optimized a_i

orbital positions a_i (red dots) for a benzene molecule before (lhs) and after their optimization (rhs). The Fermi orbitals can be interpreted as bond orbitals. A e advantage of the Fermi orbitals as compared to the Kohn-Sham-orbitals lies in the possibility of providing for a physical or chemical interpretation.

Results - second order derivative

gonal elements of the second order derivative were calculated analytically and implemented in a version of NRLMOL. Explicitly, they read

$$\frac{\mathrm{d}^2 E^{\mathsf{SIC}}}{\mathrm{d}a_m^2} = -2\sum_{jk} \left\langle \phi_j \Big| V_k^{\mathsf{SIC}} \Big| \phi_k \right\rangle \left\langle \frac{\mathrm{d}\phi_k}{\mathrm{d}a_m} \Big| \frac{\mathrm{d}\phi_j}{\mathrm{d}a_m} \right\rangle + 2\sum_{jkl} \left\langle \phi_j \Big| V_k^{\mathsf{SIC}} \Big| \phi_l \right\rangle \left\langle \frac{\mathrm{d}\phi_k}{\mathrm{d}a_m} \Big| \phi_j \right\rangle \left\langle \phi_l \Big| \frac{\mathrm{d}\phi_k}{\mathrm{d}a_m} \right\rangle \ .$$

nis a significant improvement of the self-interaction correction can be achieved (Poster MM 15.17).

J. Perdew, A. Zunger, Phys. Rev. B 23, 5048 (1981)
 M. R. Pederson et al., J. Chem. Phys., vol. 140, 121103 (2014)
 M. D. D. J. L. L. Chem. Phys. J 40, 064112 (2015)

Institute of Theoretical Physics TU Bergakademie Freiberg Contact: Charlotte Vogelbusch charlotte vogelbusch@student tu-freiberg de Fermi-orbital based self-interaction corrected DFT

$$E_{\sigma}^{\chi} = -\frac{1}{2} \int d^{3}\boldsymbol{r} \rho_{\sigma}(\boldsymbol{r}) \int d^{3}\boldsymbol{a} \Big\{ \frac{\rho_{\sigma}(\boldsymbol{r}, \boldsymbol{a}) \rho_{\sigma}(\boldsymbol{a}, \boldsymbol{r})}{\sqrt{\rho_{\sigma}(\boldsymbol{r})} \sqrt{\rho_{\sigma}(\boldsymbol{a})}} \Big\} \frac{1}{|\boldsymbol{r} - \boldsymbol{a}|}$$

- Perdew 1995: exchange-hole has properties of a single-orbital density $F_{i\sigma}(\mathbf{r}) = \frac{\sum_{\alpha} \psi_{\alpha\sigma}^*(\mathbf{a}_{i\sigma})\psi_{\alpha\sigma}(\mathbf{r})}{\sqrt{\sum_{\alpha} |\psi_{\alpha\sigma}(\mathbf{a}_{i\sigma})|^2}}$
- The Fermi-orbital yields that single-orbital density!
- Fermi-orbitals (Luken and Beratan 1982,Luken and Culberson, 1984) are a special set of localized orbitals
- At r = a_{iσ}, the square of the FO equals to the total spin density at that point