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Combining Wavefunction Theory with SCAN to Describe Electron Correlation

Alejandro J. Garza

Department of Chemistry, Rice University, Houston, TX

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| Introduction | |
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First of all, a few useful definitions:

Electron Correlation: The difference between the exact and HF energies.

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First of all, a few useful definitions:

- **D** *Electron Correlation:* The difference between the exact and HF energies. Can be classified in three types:
 - *Dynamic or short-range weak*, due to instantaneous electron repulsions.
 - *Dispersion or long-range weak*, from instantaneous multipole-multipole interactions between electrons on different centers.
 - *Static or strong*, which occurs due to degeneracies.

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Taking into account correlation is crucial for describing many physical and chemical phenomena.

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The dissociation of H_2 is the paradigm for the division of static and dynamic correlation



Figure 1: Dissociation energy curves for the H₂ molecule

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The dissociation of H_2 is the paradigm for the division of static and dynamic correlation



Figure 2: Dissociation energy curves for the H₂ molecule

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Static and dynamic correlation are pervasive in bond breaking processes



Figure 3: Dissociation energy curves for the N_2 molecule.

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Static and dynamic correlation are pervasive in bond breaking processes



Figure 4: Dissociation energy curves for the N_2 molecule.

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Static and dynamic correlation are pervasive in bond breaking processes



Figure 5: Dissociation energy curves for the N₂ molecule.

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The correlation lost by multireference methods may be critical for weak interactions



CCD0 sacrifices some weak correlation in order to describe strong correlation. However, the missing correlation may be very important in case of more subtle interactions.



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Roughly, we can say the following about contemporary quantum chemistry methods:

- Good approximations exist for the short-range dynamic correlation (e.g., CC and DFT).
- ② Dispersion correlation is more difficult, although methods like CC and the RPA are adequate.
- 3 However, practical methods for static and dynamic correlation are still out of reach.

We attempt to tackle this problem via combinations of wavefunction and density functional theories



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It seems logical to try to combine WFN methods with DFT to tackle the correlation problem

Roughly, we can say

 $WFN \rightarrow Can handle static correlation$

 $DFT \rightarrow Efficiently \ captures \ dynamic \ correlation$

Thus, we may expect a good approximation to the exact energy to be



However, such combinations must address three problems:

- **1** WFN Method Problems
- 2 The "Symmetry Dilemma"
- 3 Double Counting

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The problems of standard WFN+DFT can be addressed by a combination of CCD0 with SCAN

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SPECIAL ISSUE IN HONOUR OF ANDREAS SAVIN

Combinations of coupled cluster, density functionals, and the random phase approximation for describing static and dynamic correlation, and van der Waals interactions

Alejandro J. Garza^a, Ireneusz W. Bulik^a, Ana G. Sousa Alencar^a, Jianwei Sun^b, John P. Perdew^{b,c} and Gustavo E. Scuseria^{a,d}

^aDepartment of Chemistry, Rice University, Houston, TX, USA; ^aDepartment of Physics, Temple University, Philadelphia, PA, USA; ^cDepartment of Chemistry, Temple University, Philadelphia, PA, USA; ^cDepartment of Physics and Astronomy, Rice University, Houston, TX, USA

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Actinide chemistry using singlet-paired coupled cluster and its combinations with density functionals

Alejandro J. Garza,¹ Ana G. Sousa Alencar,¹ and Gustavo E. Scuseria² ¹Department of Chemistry, Rice University, Houston, Texas 77251-1892, USA ²Department of Chemistry and Department of Physics and Astronomy, Rice University, Houston, Texas 77251-1892, USA

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CCD0 is a generalization of pCCD

The T_2 operator of standard CCD can be written as a sum of singlet- and triplet-pairing components

$$T_2 = T_2^{[0]} + T_2^{[1]}.$$

CCD0 uses only the singlet-pairing component:

$$|\Psi_{\rm CCD0}\rangle = e^{T_2^{[0]}} |\Phi_{\rm RHF}\rangle,$$

where

$$T_2^{[0]} = \frac{1}{2} \sum_{ijab} \sigma_{ij}^{ab} P_{ab}^{\dagger} P_{ij},$$

with

$$P_{ij} = \frac{1}{\sqrt{2}} \left(c_{j\uparrow} c_{i\downarrow} + c_{i\uparrow} c_{j\downarrow} \right).$$

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Avoiding double counting in CCD0+DFT

Now, looking at the full T_2 operator

$$T_2 = T_2^{[0]} + T_2^{[1]},$$

The triplet-paired component is

$$T_2^{[1]} = \frac{1}{2} \sum_{ijab} \pi_{ij}^{ab} \mathbf{Q}_{ab}^{\dagger} \cdot \mathbf{Q}_{ij}$$

where $\mathbf{Q}_{ij} = (Q_{ij}^+, Q_{ij}^0, Q_{ij}^-)^t$ with

$$Q_{ij}^{+} = c_{j\uparrow}c_{i\uparrow}, \quad Q_{ij}^{-} = c_{j\downarrow}c_{i\downarrow}, \quad Q_{ij}^{0} = \frac{1}{\sqrt{2}}\left(c_{j\uparrow}c_{i\downarrow} - c_{i\uparrow}c_{j\downarrow}\right)$$

Because the Q_{ij}^+ , Q_{ij}^0 , and Q_{ij}^- components contribute all equally to the energy we get (for a closed shell)

$$E_{\text{total}}^{\text{CCD0+DFT}} = E^{\text{CCD0}} + 3E_{c\uparrow\uparrow}^{\text{DFA}}[n].$$

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We can extract the equal spin correlation in SCAN for combination with CCD0

The SCAN functional constructs ε_c as

$$\varepsilon_c = \varepsilon_c^1 + f_c(\alpha) \left[\varepsilon_c^0 - \varepsilon_c^1 \right],$$

where $f_c(\alpha)$ is a function that depends on τ , and $\varepsilon_c^{\alpha=1}$ and $\varepsilon_c^{\alpha=0}$ are the uniform density and single orbital limits of ε_c . Hence, the parallel spin correlation density is

$$\varepsilon_c^{\uparrow\uparrow} = \varepsilon_c^{1\uparrow\uparrow} + f_c(\alpha) \left[\varepsilon_c^{0\uparrow\uparrow} - \varepsilon_c^{1\uparrow\uparrow} \right].$$

Furthermore, since $\varepsilon_c^{0\uparrow\uparrow} = 0$, we get

$$\varepsilon_c^{\uparrow\uparrow} = \varepsilon_c^{1\uparrow\uparrow} - f_c(\alpha)\varepsilon_c^{1\uparrow\uparrow},$$

and $\varepsilon_c^{1\uparrow\uparrow}$ has been parametrized by Gori–Giorgi and Perdew.¹

¹Phys. Rev. B **69**, 041103 (2004).

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CCD0+DFT is very accurate for problems dominated by dynamic correlation

Table 1: Mean absolute errors (MAEs) for various properties ofsmall molecules for which accurate reference data are available.

| | MAE | | |
|--|-------|-------|----------|
| Property | DFT | CCD0 | CCD0+DFT |
| Bond Lengths ^{\dagger} (Å) | 0.024 | 0.013 | 0.007 |
| Vibrational Frequencies [†] (cm^{-1}) | 68 | 22 | 18 |
| Proton Affinities [‡] (kcal/mol) | 2.65 | 2.69 | 0.66 |

[†]First and second row diatomics.

[‡]Standard PA8 set of proton affinities.

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Static and dynamic correlation can also be described by CCD0+DFT



Figure 7: Errors in energy for the beryllium isoelectronic series.

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However, the long-range dispersion correlation remains missing in CCD0+DFT



Figure 8: Dissociation energy curves for the He₂ dimer

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We can correct this issue by complementing with dRPA long-range correlation

The dRPA equations are:

$$\begin{pmatrix} \mathbf{A} & \mathbf{B} \\ -\mathbf{B} & -\mathbf{A} \end{pmatrix} \begin{pmatrix} \mathbf{X} & \mathbf{Y} \\ \mathbf{Y} & \mathbf{X} \end{pmatrix} = \begin{pmatrix} \mathbf{X} & \mathbf{Y} \\ \mathbf{Y} & \mathbf{X} \end{pmatrix} \begin{pmatrix} \boldsymbol{\omega} & 0 \\ 0 & -\boldsymbol{\omega} \end{pmatrix}$$

And the dRPA ground-state correlation is

$$E_c^{\mathrm{dRPA}} = \frac{1}{2} \mathrm{Tr}(\boldsymbol{\omega} - \mathbf{A}).$$

with

$$\begin{split} A_{ia,jb} &= (\epsilon_a - \epsilon_i) \delta_{ij} \delta_{ab} + \langle ib | aj \rangle, \\ B_{ia,jb} &= \langle ij | ab \rangle. \end{split}$$

So we can take only same spin integrals to get same spin correlation, and evaluate them only in the long range.

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DFT and the dRPA are good, complementary choices for evaluating residual correlation



Separation of electron-electron interaction into short-Figure 9: and long-ranges.

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Therefore, we have the following approximations:

1 CCD0+DFT: For problems in which long range interactions are not important

$$E = \underbrace{E_{\text{CCD0}}}_{\mathscr{O}(N^6)} + \frac{3}{2} \underbrace{\left(E_{c \uparrow \uparrow}^{\text{DFT}} + E_{c \downarrow \downarrow}^{\text{DFT}} \right)}_{\mathscr{O}(N^3)}.$$

Long-range corrected (LC)-CCD0+DFT: For problems involving long range interactions

$$E = \underbrace{E_{\text{CCD0}}}_{\mathscr{O}(N^6)} + \frac{3}{2} \underbrace{\left(E_{c\uparrow\uparrow}^{\text{lr-dRPA}} + E_{c\downarrow\downarrow}^{\text{lr-dRPA}} \right)}_{\approx \mathscr{O}(N^4) - \mathscr{O}(N^5)} + \frac{3}{2} \underbrace{\left(E_{c\uparrow\uparrow}^{\text{sr-DFT}} + E_{c\downarrow\downarrow}^{\text{sr-DFT}} \right)}_{\mathscr{O}(N^3)}.$$

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Long-range dispersion can now be described with LC-CCD0+DFT



Figure 10: Dissociation energy curves for the He₂ dimer

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The different CCD0+DFT methods can also describe challenging bond breaking processes



Figure 11: Dissociation energy profiles for the N_2 molecule using a cc-pVDZ basis.

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A challenging system: Beryllium dimer—an "impossible molecule"



Figure 12: Dissociation energy profiles for the Be₂ dimer using an uncontracted aug-cc-pVTZ basis.

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Application of CCD0+DFT to actinide chemistry: the uranyl ion UO_2^{2+}

Table 2: Bond lengths $(R_e \text{ in } \text{\AA})$ and harmonic vibrational frequencies $(\omega_e \text{ in } \text{cm}^{-1})$ for UO_2^{2+} .

| Method | R_e | ω_{as} | $\omega_{ m s}$ | ω_{eta} |
|---------------|-------|------------------------|-----------------|----------------|
| HF | 1.648 | 1293 | 1220 | 267 |
| DFT | 1.715 | 1086 | 985 | 123 |
| CCD0 | 1.698 | 1129 | 1053 | 541 |
| CCD0+DFT | 1.693 | 1121 | 1052 | 220 |
| Best Estimate | 1.690 | 1120 | 1035 | 178 |

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Conclusions •0

Conclusions for CCD0+DFT:

Advantages

- Avoids the usual problems of standard WFN+DFT.
- 2 No need of empirical parameters or symmetry breaking.
- 3 Reasonable description of static and dynamic correlation.
- Can also capture dispersion correlation in its LC-CDD0+DFT variant.

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Conclusions for CCD0+DFT:

Advantages

- Avoids the usual problems of standard WFN+DFT.
- 2 No need of empirical parameters or symmetry breaking.
- 3 Reasonable description of static and dynamic correlation.
- Can also capture dispersion correlation in its LC-CDD0+DFT variant.

Disadvantages

Same computational cost as coupled cluster.

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- 4 Dr. Ireneusz Bulik
- 5 Dr. Thomas Henderson
- 6 Ana G. Sousa Alencar

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