#### X-ray crystallography tutorial

CCDM
Mike Zdilla
Temple Chemistry

#### Nobel Prize winners associated with crystallography

2013 Chemistry

M. Karplus, M. Levitt and A. Warshel

For the development of multiscale models for complex chemical systems

2012 Chemistry

R. J. Lefkowitz and B. K. Kobilka

For studies of G-protein-coupled receptors

2011 Chemistry

D. Shechtman

For the discovery of quasicrystals 2010 Physics

A. Geim and K. Novoselov

For groundbreaking experiments regarding the two-dimensional material graphene

2009 Chemistry

V. Ramakrishnan, T. A. Steitz and A. E. Yonath

Studies of the structure and function of the ribosome 2006 Chemistry

R. D. Kornberg

Studies of the molecular basis of eukaryotic transcription

2003 Chemistry

R. MacKinnon

Potassium channels

1997 Chemistry

P. D. Boyer, J. E. Walker and J. C. Skou

Elucidation of the enzymatic mechanism underlying the synthesis of adenosine triphosphate (ATP) and

discovery of an ion-transporting enzyme

1996 Chemistry

R.Curl, H. Kroto and R. Smalley

Discovery of the fullerene form of carbon

1994 Physics

C. Shull and N. Brockhouse

Neutron diffraction

1992 Physics

G. Charpak

Discovery of the multi wire proportional chamber

1991 Physics

P.-G. de Gennes

Methods of discovering order in simple systems can be applied to polymers and liquid crystals

1988 Chemistry

J. Deisenhofer, R. Huber and H. Michel

For the determination of the three-dimensional structure of a photosynthetic reaction centre

1985 Chemistry

H. Hauptman and J. Karle

Development of direct methods for the determination of crystal structures

1982 Chemistry

A. Klug

Development of crystallographic electron microscopy and discovery of the structure of biologically important nucleic acid-protein complexes

1976 Chemistry

W. N. Lipscomb

Structure of boranes

1972 Chemistry

C. B. Anfinsen

Folding of protein chains

1964 Chemistry

D. Hodgkin

Structure of many biochemical substances including Vitamin B12

1962 Physiology or Medicine

F. Crick, J. Watson and M. Wilkins

The helical structure of DNA

1962 Chemistry

J. C. Kendrew and M. Perutz

For their studies of the structures of globular proteins

1954 Chemistry

L. C. Pauling

For his research into the nature of the chemical bond and its application to the elucidation of the structure of complex substances

1946 Chemistry

J. B. Sumner

For his discovery that enzymes can be crystallised

1937 Physics

C. J. Davisson and G. Thompson

Diffraction of electrons by crystals

1936 Chemistry

P. J. W. Debye

For his contributions to our knowledge of molecular structure through his investigations on dipole

 $moments \ and \ on \ the \ diffraction \ of \ X-rays \ and \ electrons \ in \ gases$ 

1929 Physics

L.-V. de Broglie

The wave nature of the electron

1917 Physics

C. G. Barkla

Discovery of the characteristic Röntgen radiation of the elements 1915 Physics

W. H. Bragg and W. L. Bragg

Use of X-rays to determine crystal structure

1914 Physics

M. Von Laue

Diffraction of X-rays by crystals

1901 Physics

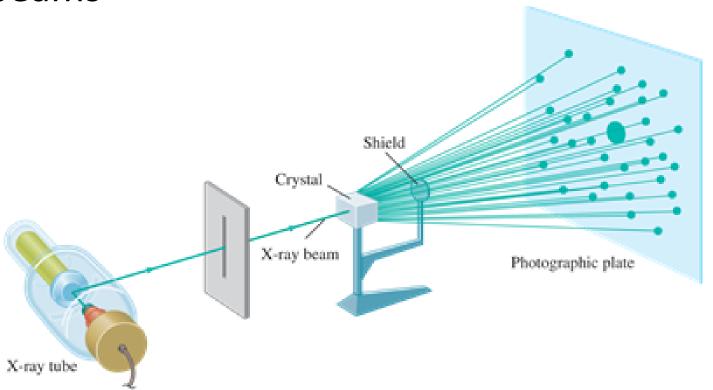
W. C. Röntgen

Discovery of X-rays

Source: International Union of Crystallography (http://www.iucr.org/people/nobel-prize)

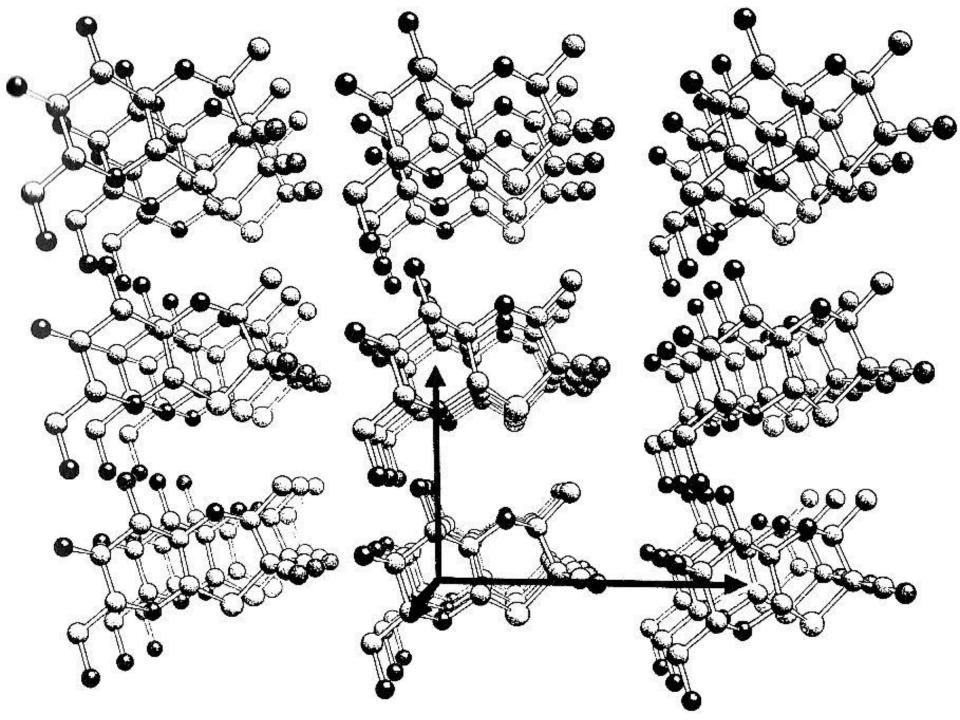
### X-ray diffraction on crystals

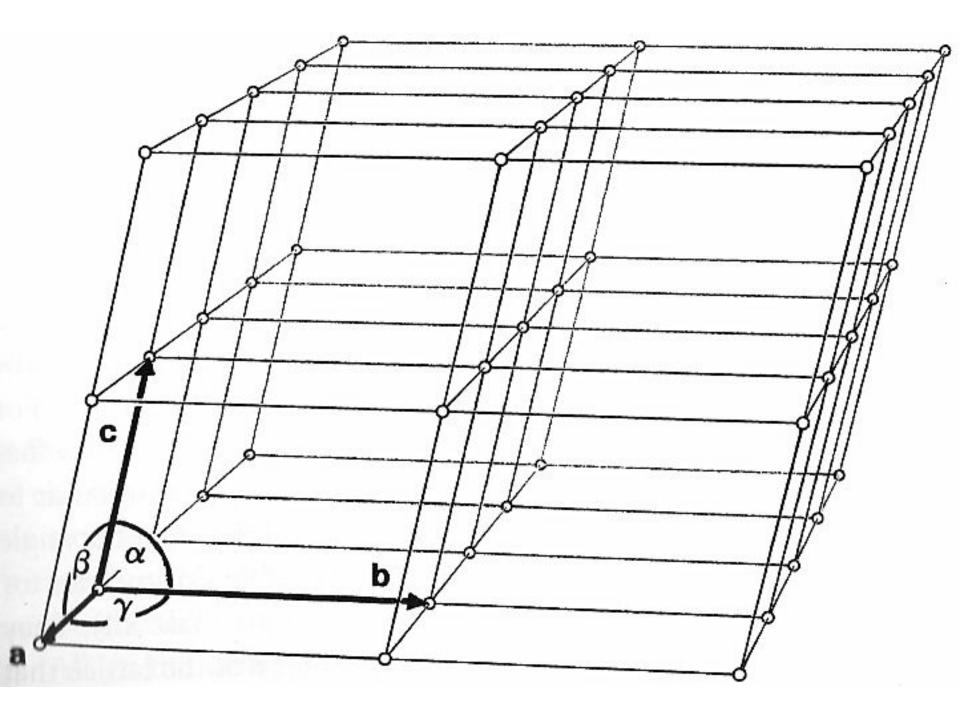
 A monochromatized beam of X-rays diffracts off of atoms in a crystal, giving new diffracted beams

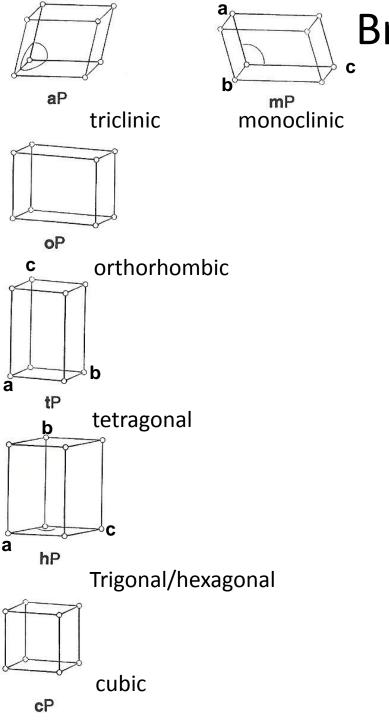


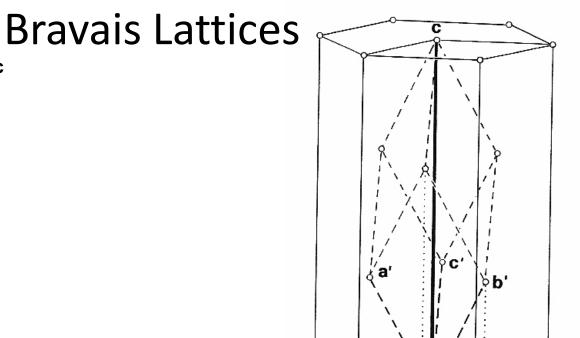
#### Diffraction experiments

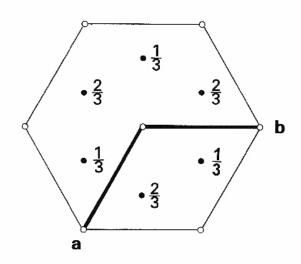
- X-ray
  - Single-crystal
  - powder
- Electron
- Neutron





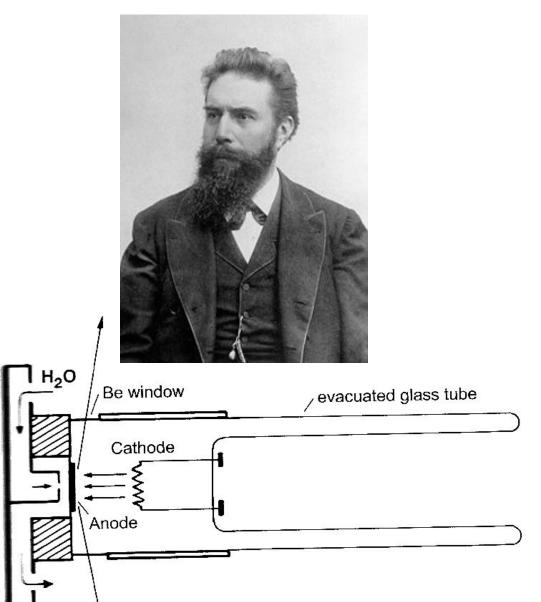




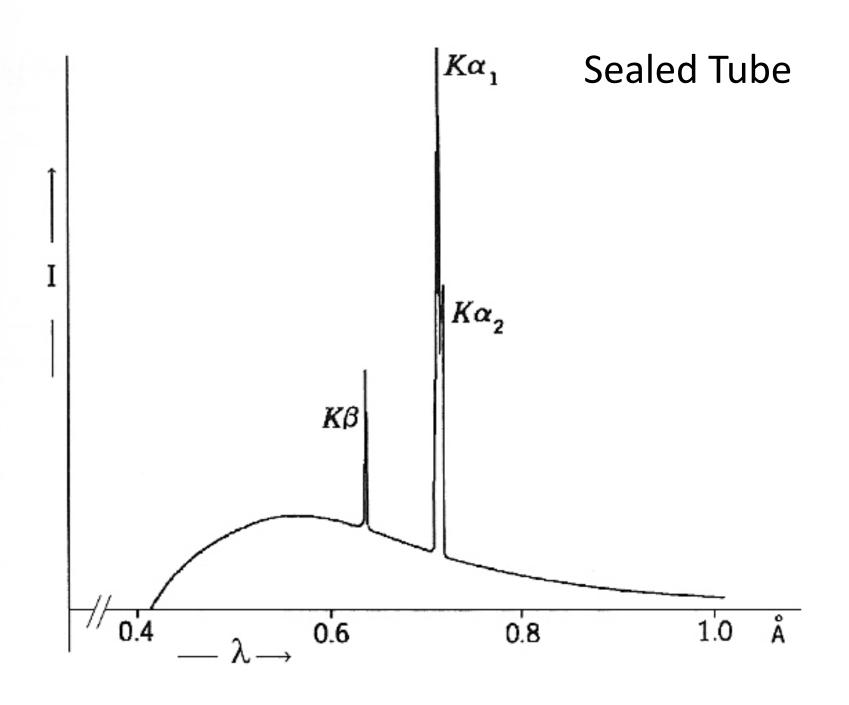


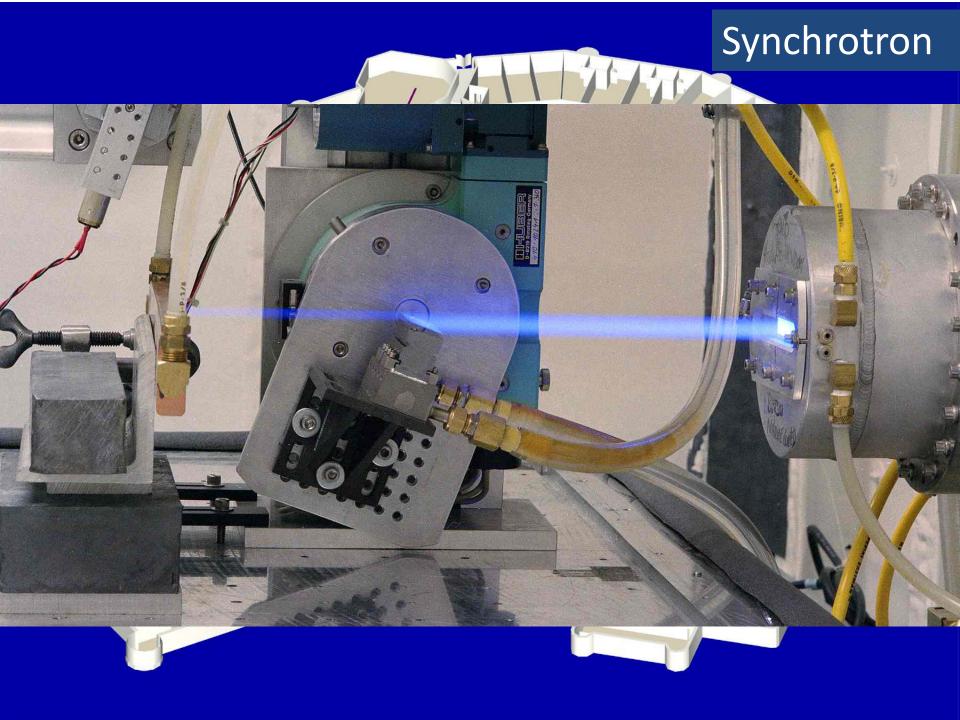
#### Wilhelm Röntgen

Discoverer of X-rays

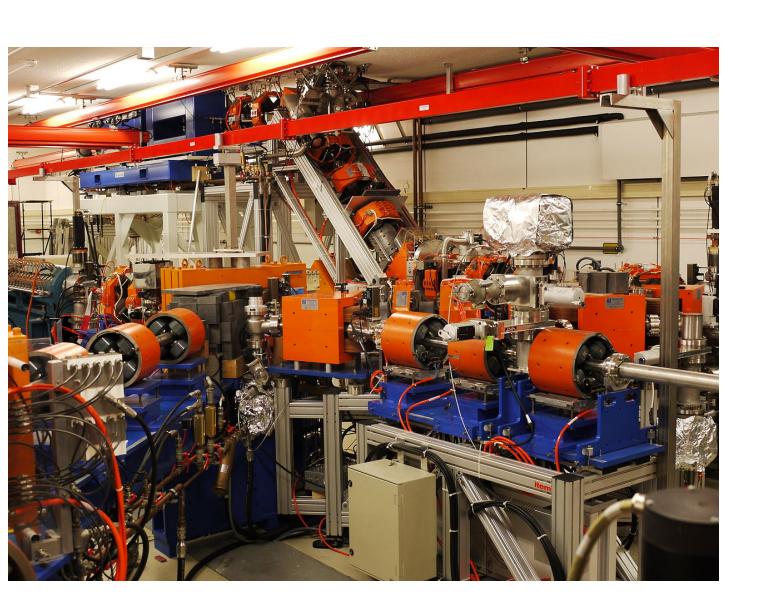




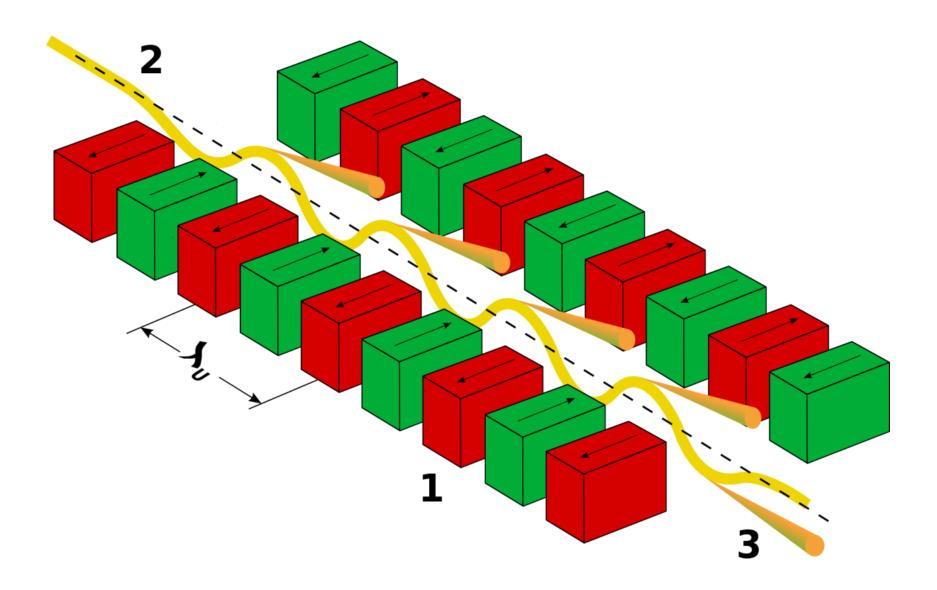




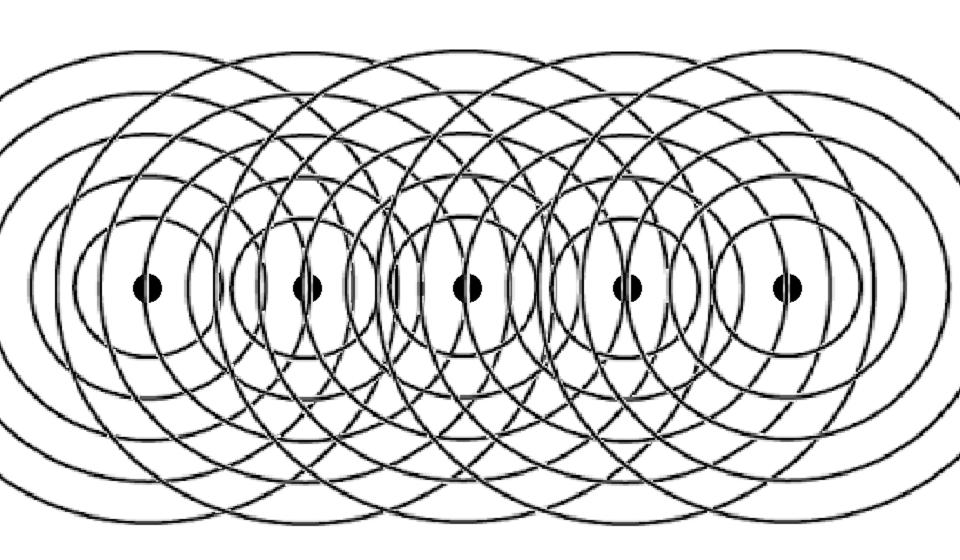
## X-ray Free Electron Laser (XFEL)



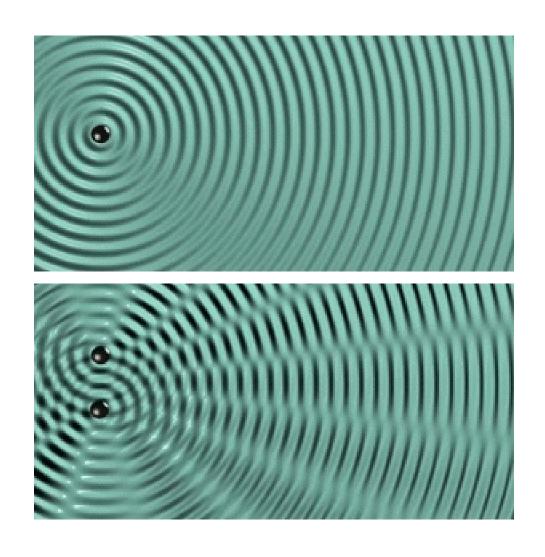
Electrons Oscilate as they travel through alternating magnetic fields.

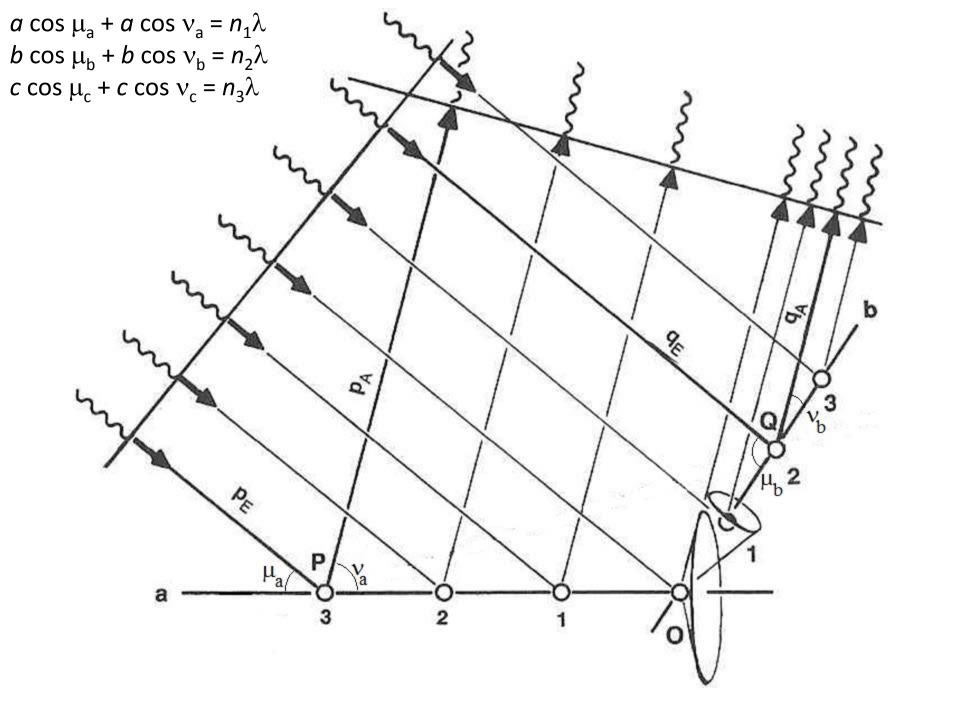


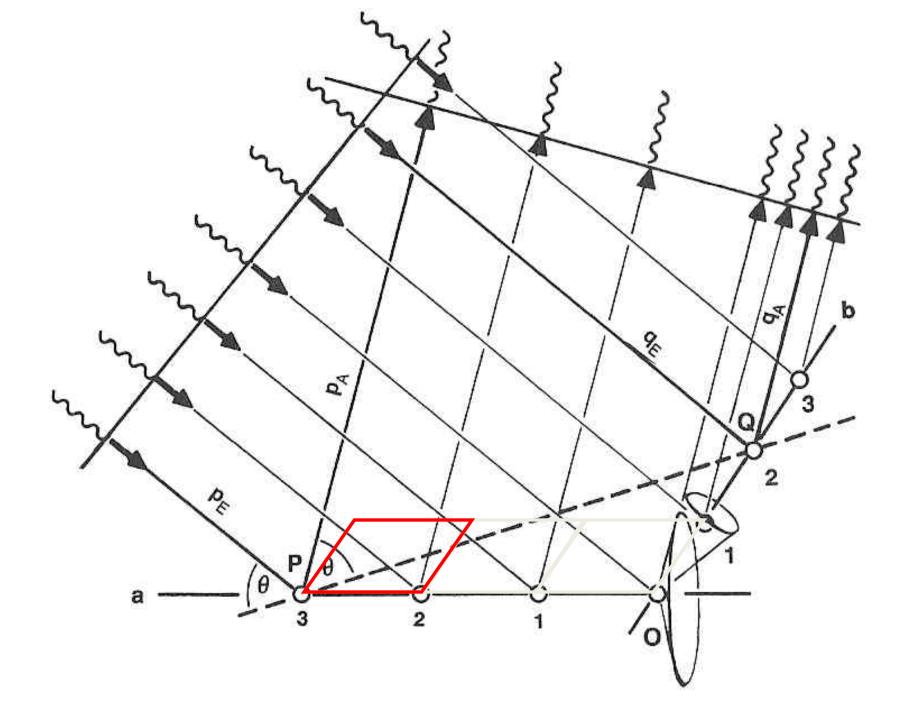
## Scattering



## Single vs. Multi-atom scattering

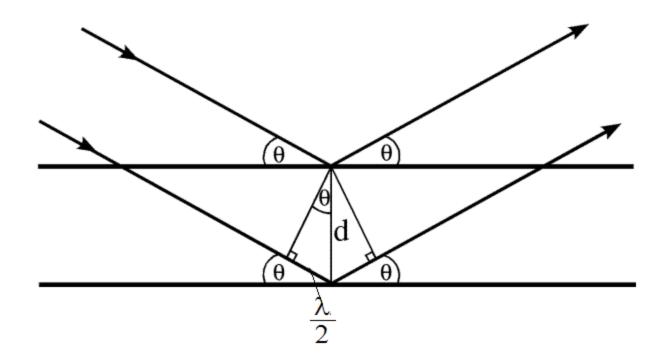




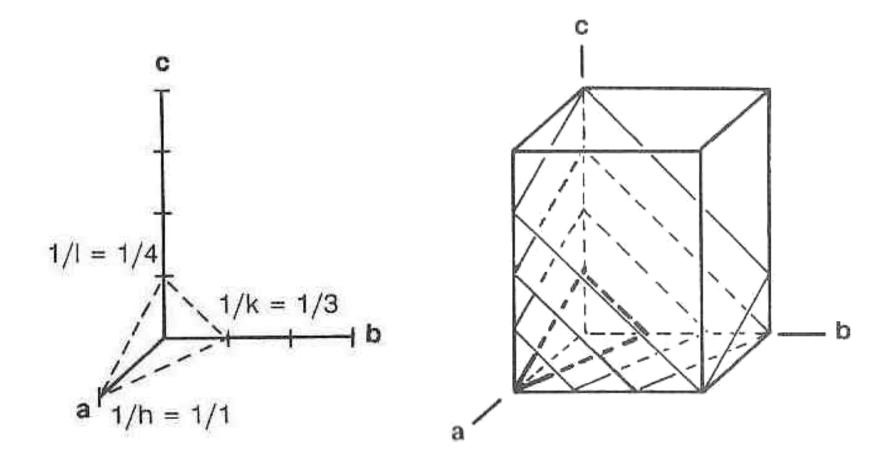


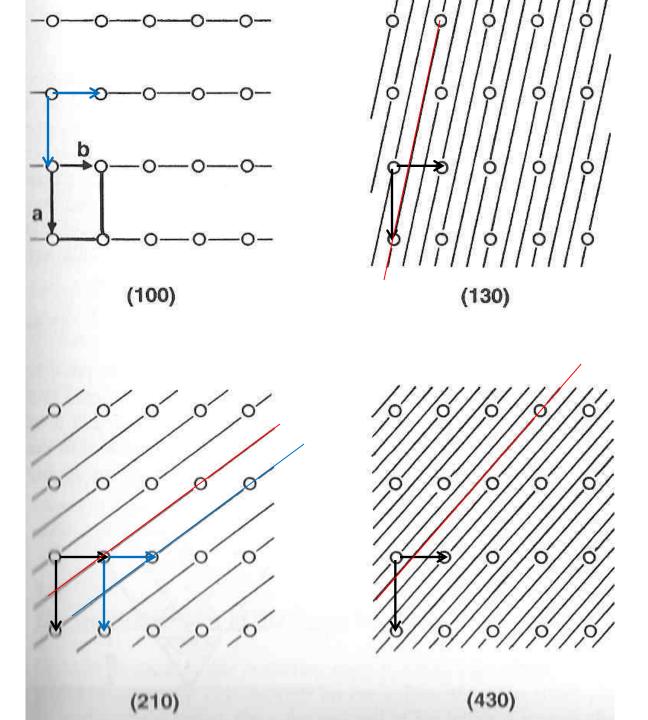
#### Bragg equation

 $\sin \theta = n\lambda/2d$   $2d \sin \theta = n\lambda$ 

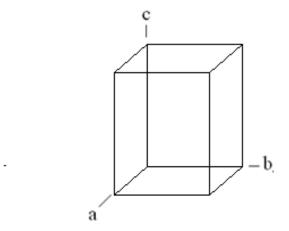


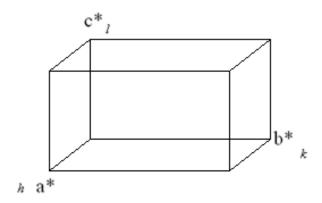
## hkl = 134





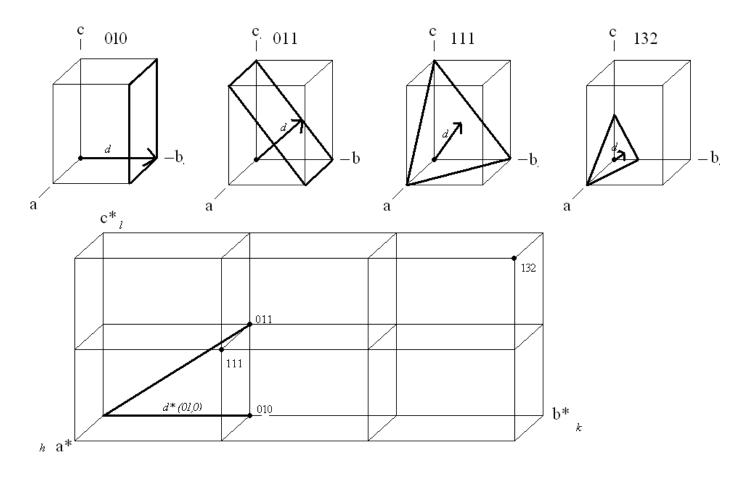
#### Reciprocal Lattice





$$a^* = \frac{b \times c}{V} b^* = \frac{a \times c}{V} c^* = \frac{a \times b}{V}$$
 $a^* = 1/a, b^* = 1/b, c^* = 1/c$ 

## Reciprocal Lattice



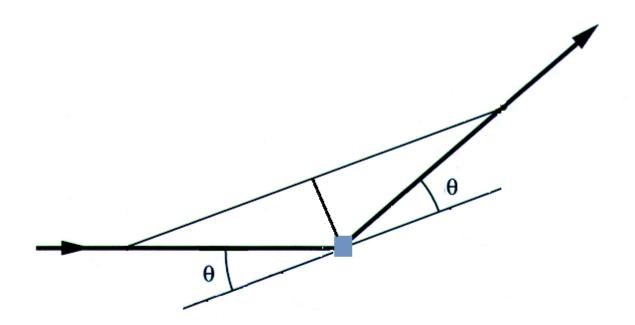
• 
$$d^* = 1/d$$
,  $a^* = 1/a$ ,  $b^* = 1/b$ ,  $c^* = 1/c$ 

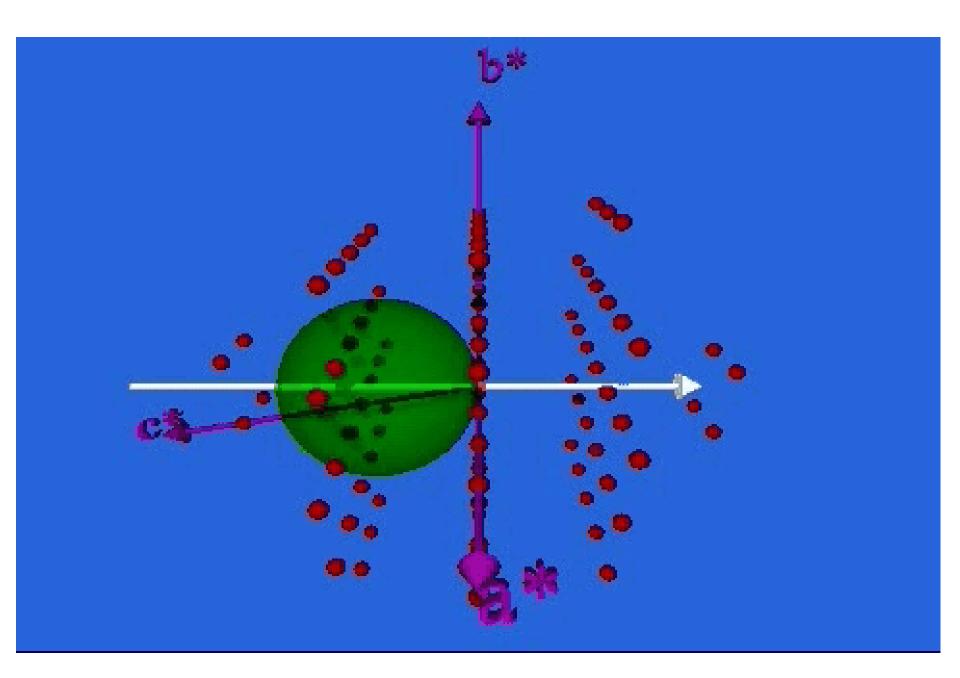
• 
$$d^{*2} = h^2 a^{*2} + k^2 b^{*2} + l^{*2} c^2$$

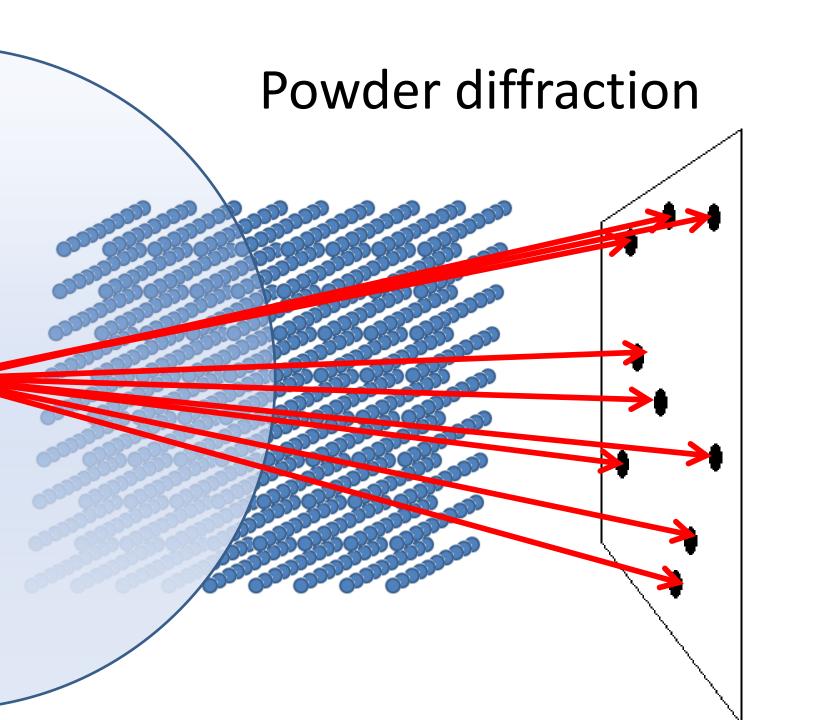
• 
$$r^2 = x^2 + y^2 + z^2$$

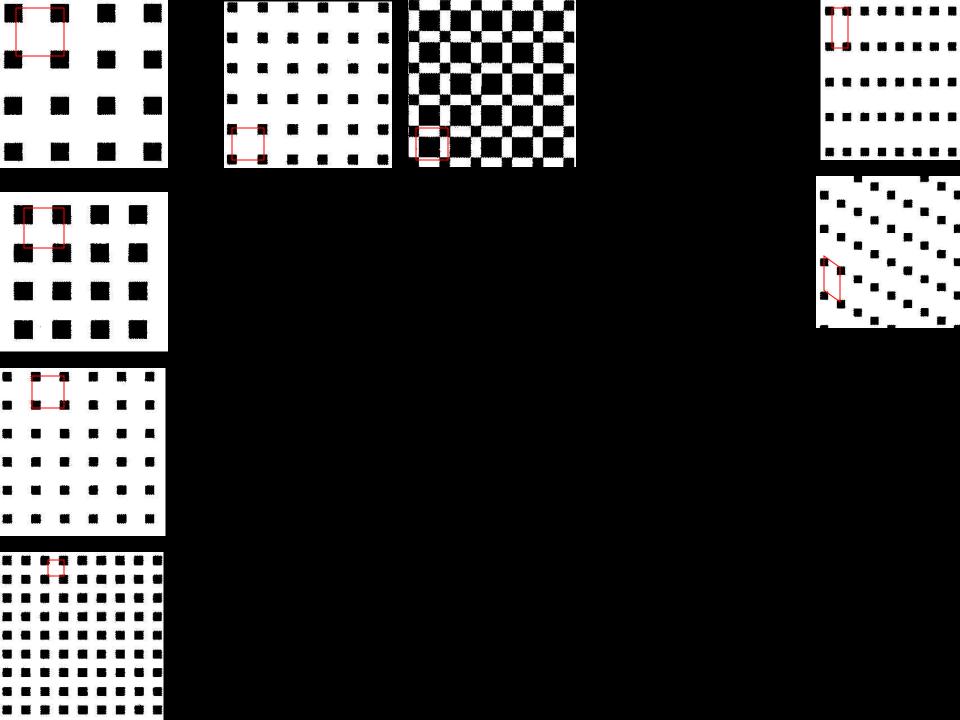
# Ewald Sphere and the Reciprocal Lattice

 $\sin \theta = \lambda/2d$  $2d \sin \theta = \lambda$ 

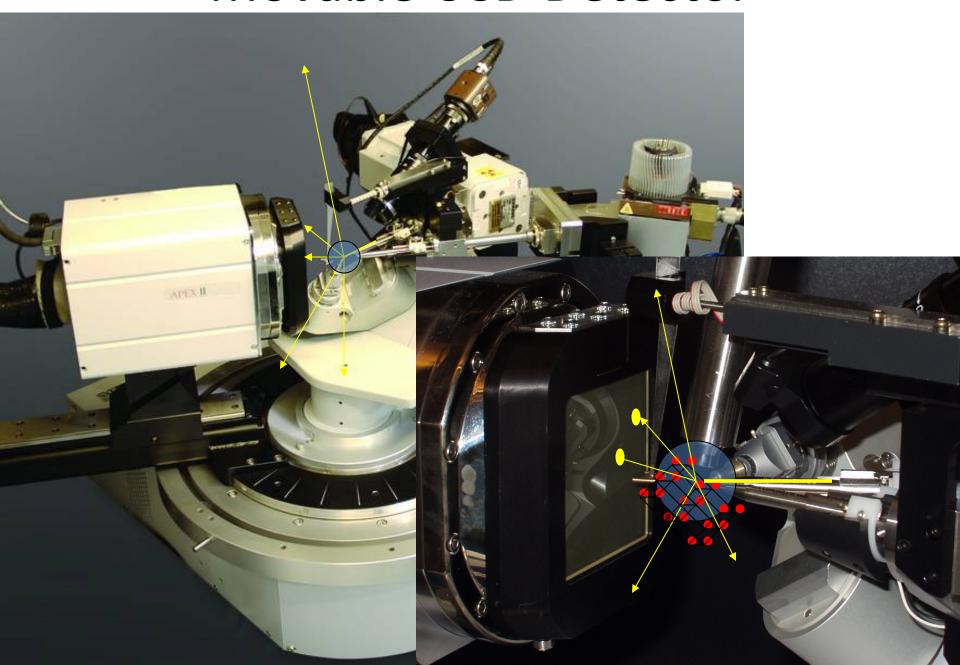


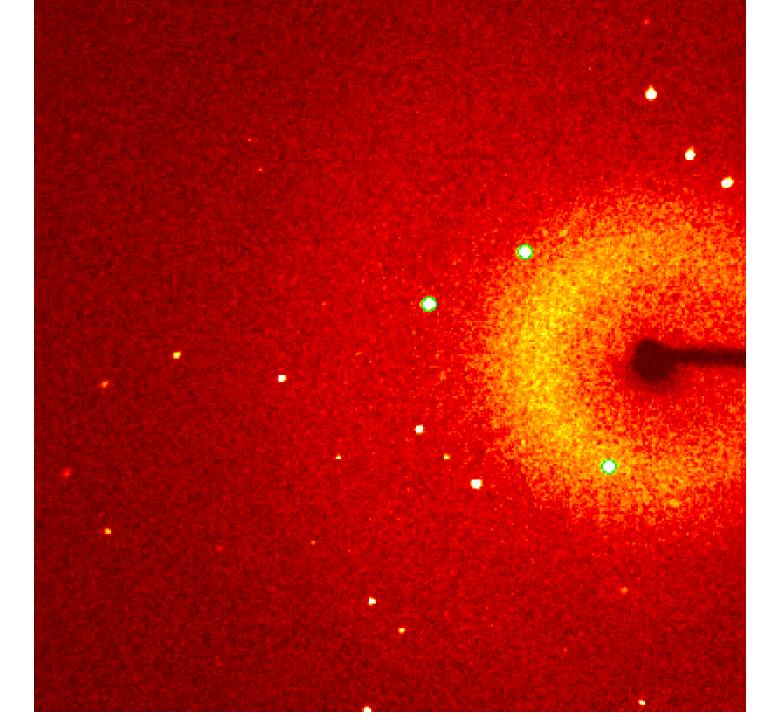


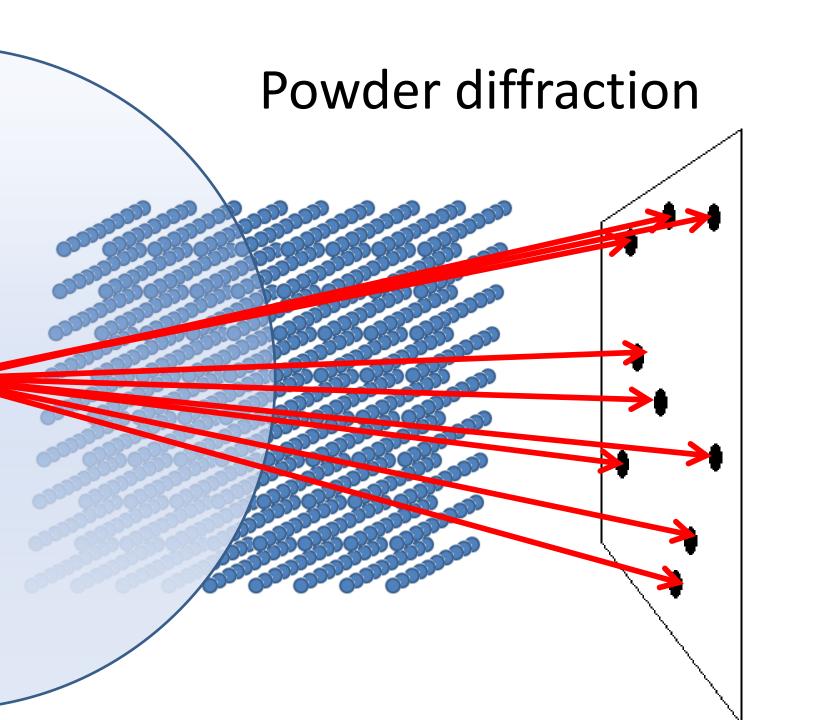


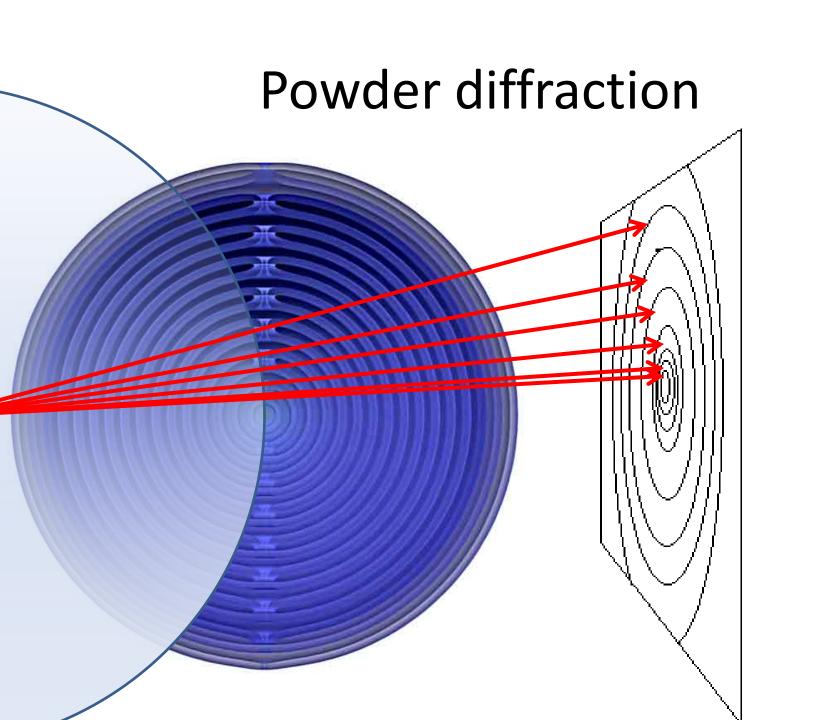


#### Movable CCD Detector

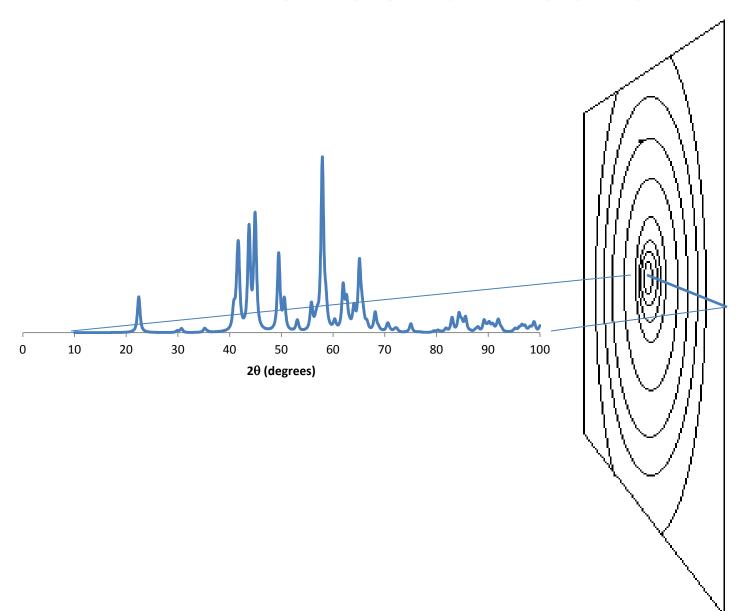








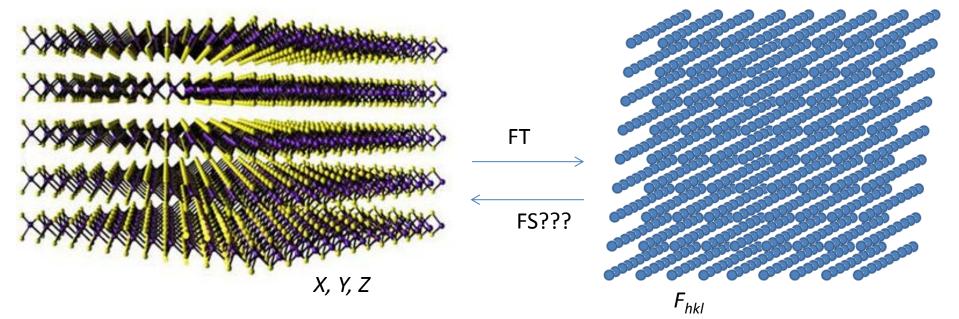
#### Powder diffraction



#### Solving the crystal structure:

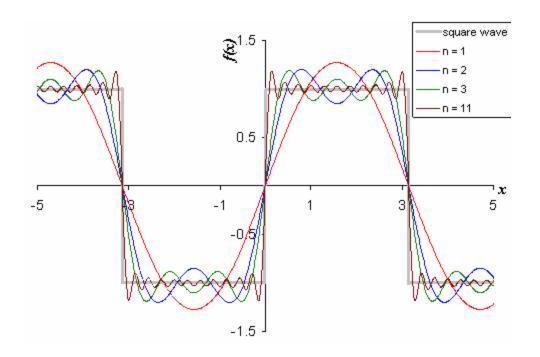
 Relationship between the intensity weighted reciprocal lattice and the atomistic structure is a fourier transform

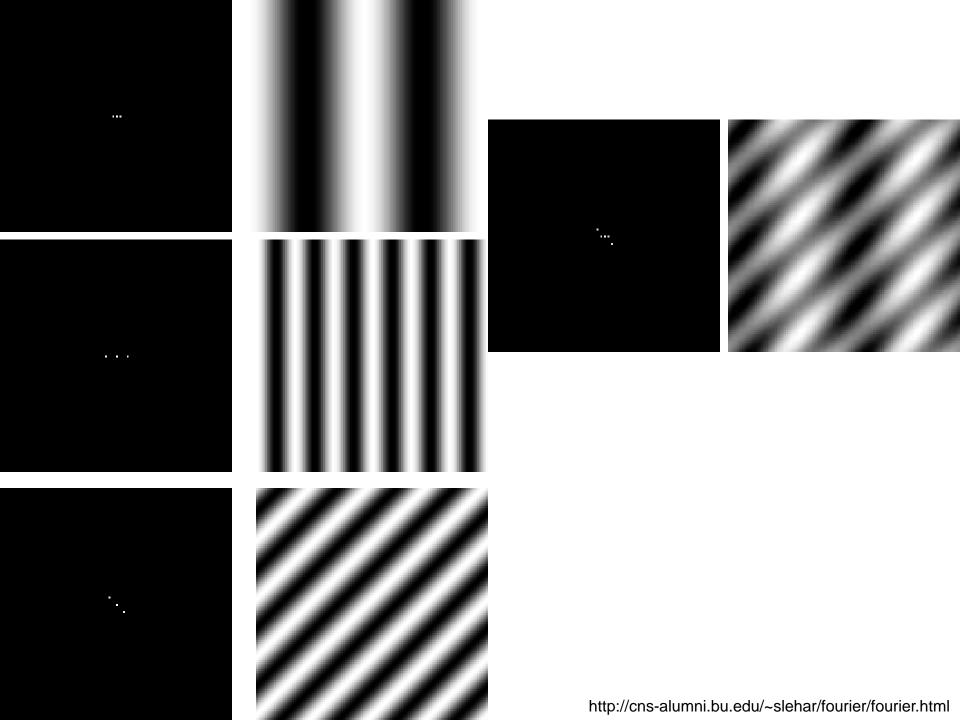
$$F_{hkl} = \sum f_i e^{i2\pi(hxi + kyi + lzi)}$$



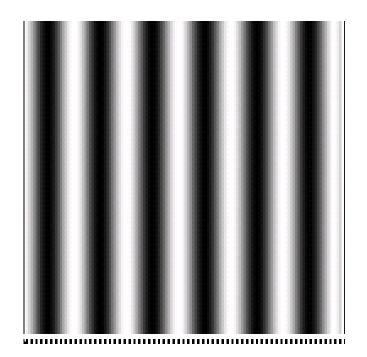
#### Fourier synthesis

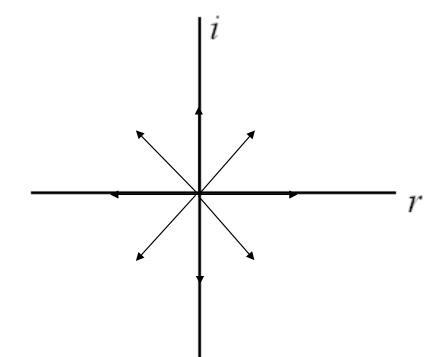
 You can construct any periodic function by the addition of sin waves, properly weighted, and properly phased.



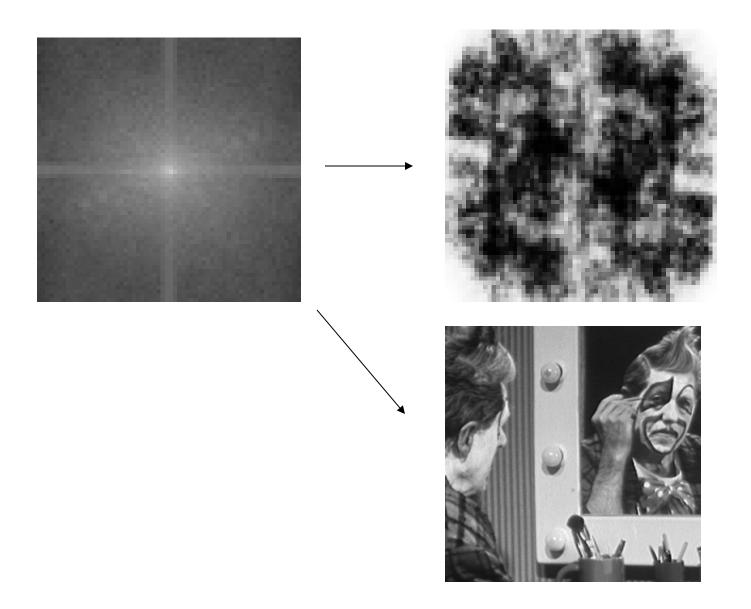




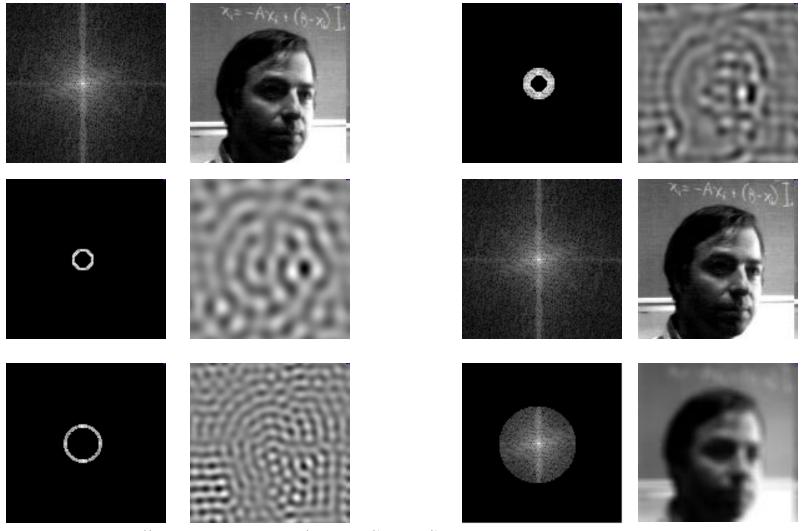




#### http://homepages.inf.ed.ac.uk/rbf/HIPR2/fourier.htm



#### Fourier synthesis of "this guy"

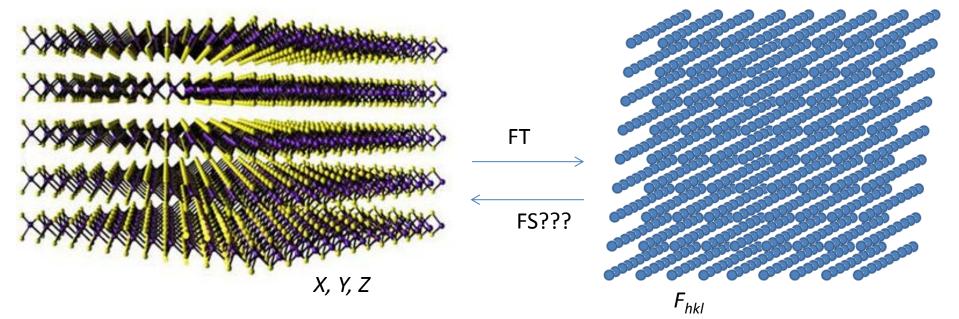


http://cns-alumni.bu.edu/~slehar/fourier/fourier.html

#### Solving the crystal structure:

 Relationship between the intensity weighted reciprocal lattice and the atomistic structure is a fourier transform

$$F_{hkl} = \sum f_i e^{i2\pi(hxi + kyi + lzi)}$$



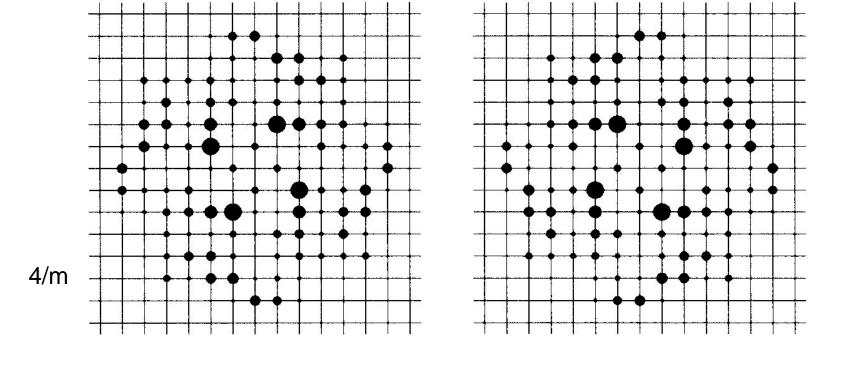
#### How to solve the structure:

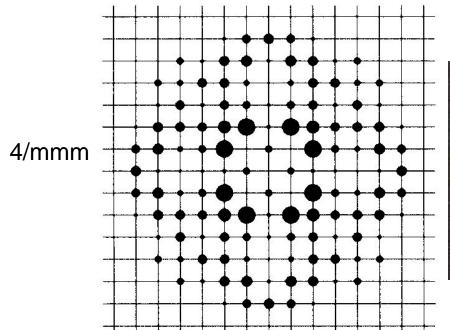
- Patterson/heavy atom methods:
  - A FS with all phases = 0 gives information from which the heavy atoms alone can be located.
  - Phases approximated from heavy atom locations
  - SF with approximated phases gives remaining atoms
  - Least squares refinement of atomic positions against diffraction data.
- Direct methods
  - Karle and Hauptman, Nobel Prize, 1985
  - Rigorous mathematical relationships between spot intensities and phases.
  - Computationally intensive

## Recent and Current Frontiers in Crystallography

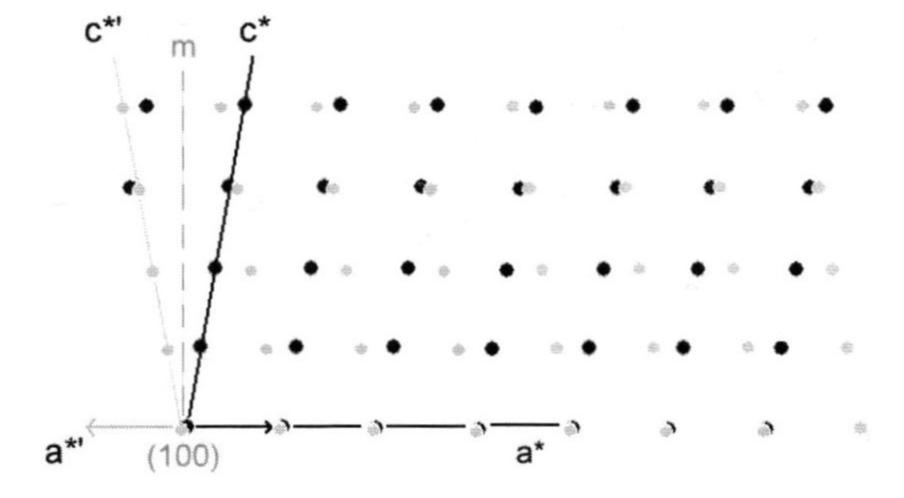
- Twinning
  - Prevents the direct obtainment of the intensity weighted reciprocal lattice.

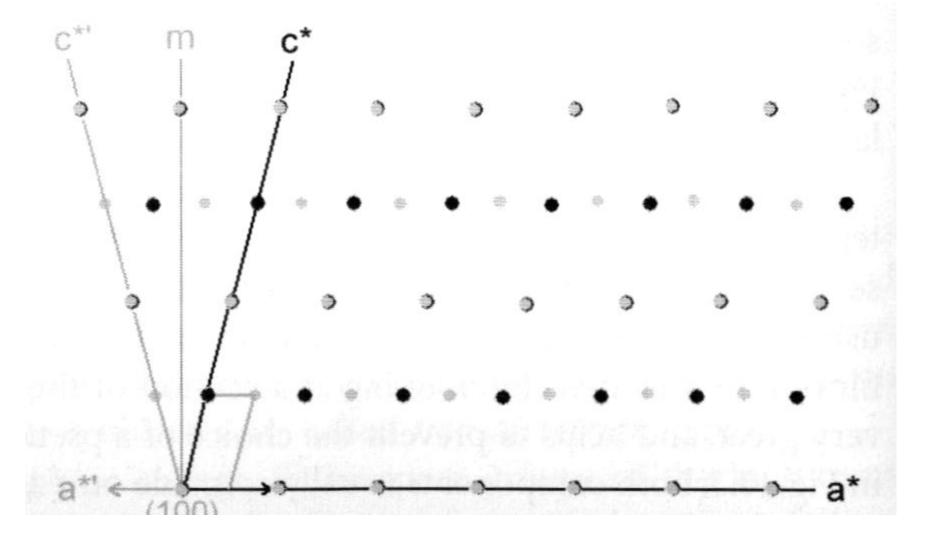








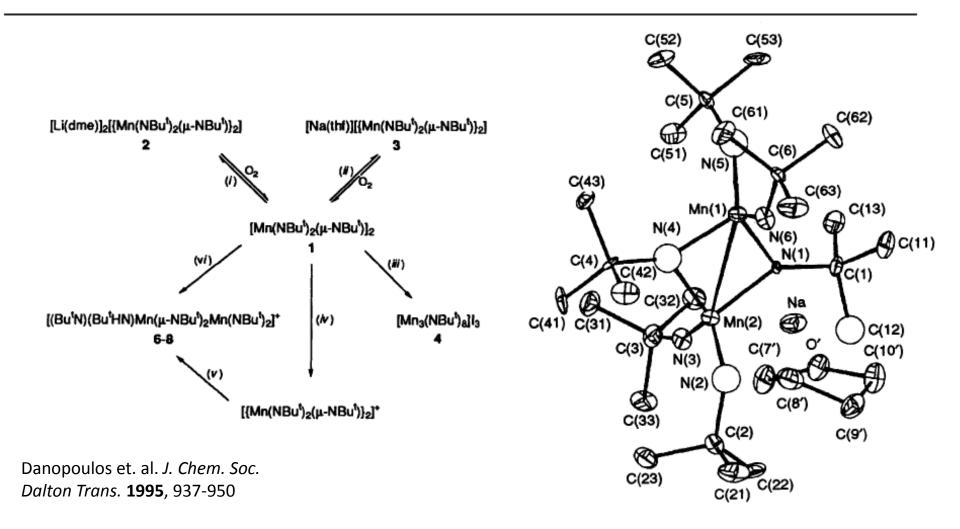




#### Anionic and Cationic Bi- and Tri-nuclear tert-Butylimido Complexes of Manganese-(v) and -(vı)†

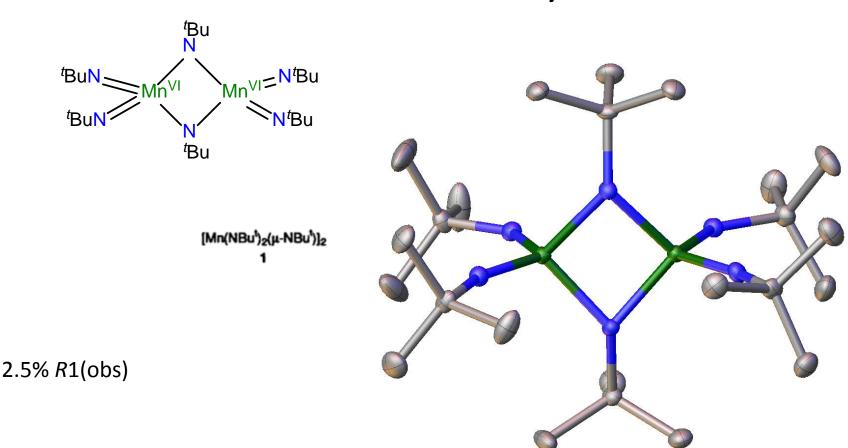
Andreas A. Danopoulos,<sup>a</sup> Geoffrey Wilkinson,\*,<sup>a</sup> Tracy K. N. Sweet<sup>b</sup> and Michael B. Hursthouse\*,<sup>b</sup>

b Department of Chemistry, University of Wales Cardiff, PO Box 912, Cardiff CF1 3TB, UK

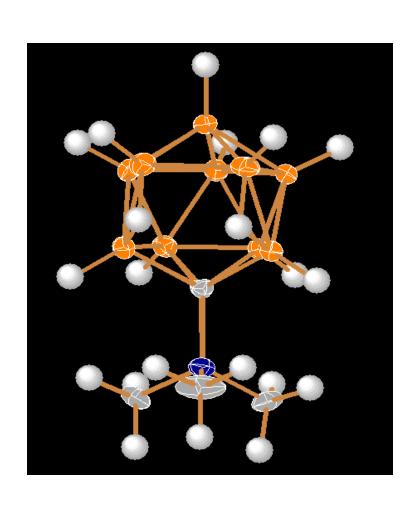


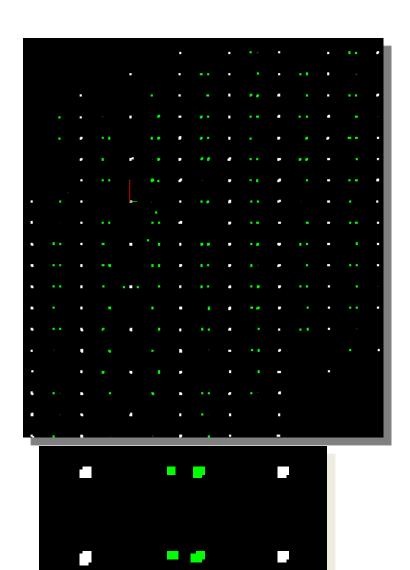
<sup>&</sup>lt;sup>a</sup> Johnson Matthey Laboratory, Chemistry Department, Imperial College, London SW7 2AY, UK

# Structure solution of a Mn(VI) dimer indexed using CELL\_NOW (George Sheldrick)

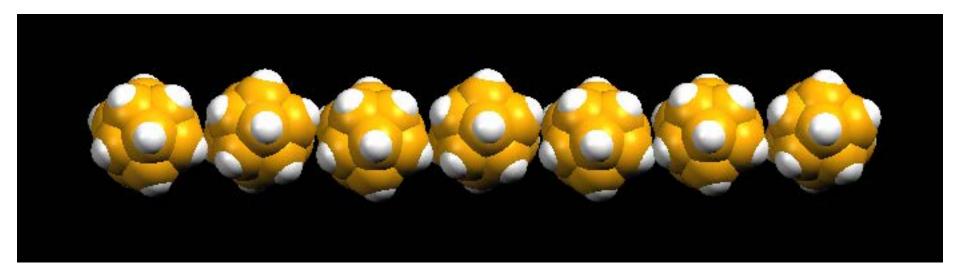


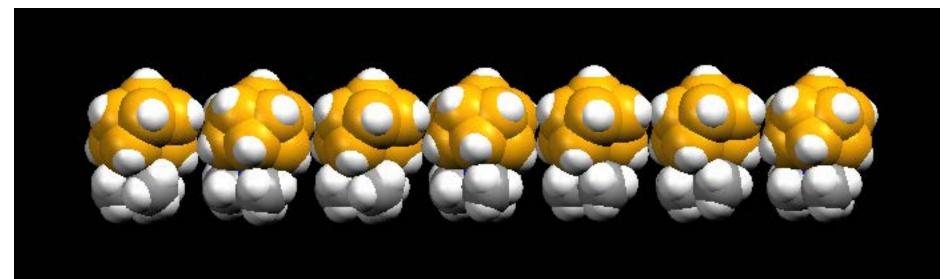
#### Modulated Carborane (Chuck Campana)



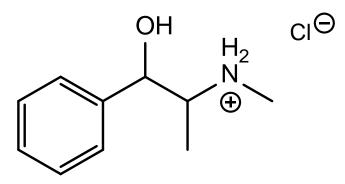


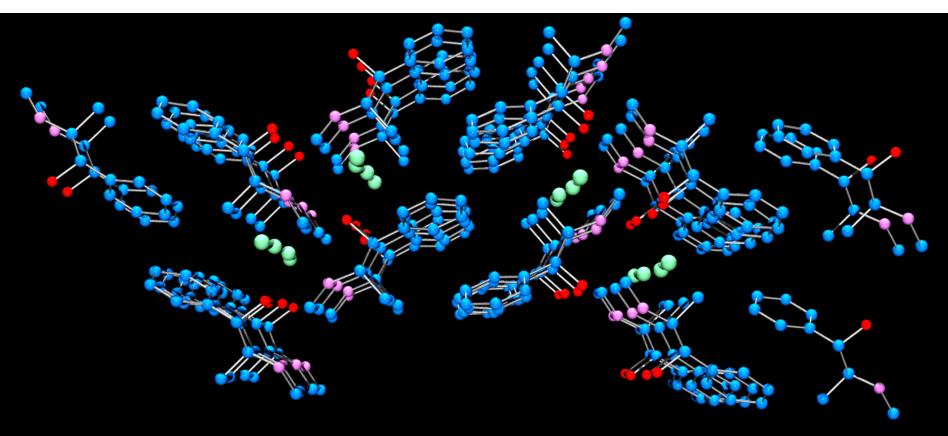
## Supercell of Commensurate Modulated structure



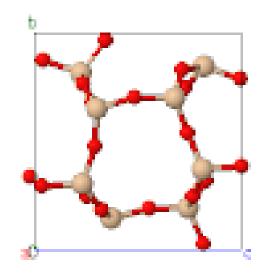


#### Ephedrine · HCl





### Incommensurate modulated structures: Tridymite, hightemperature SiO<sub>2</sub>

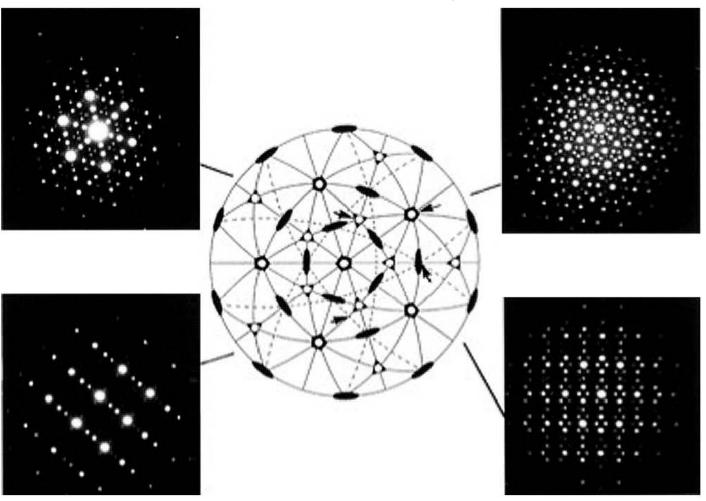


Acta Cryst. B. 2009, 65, 543-550.

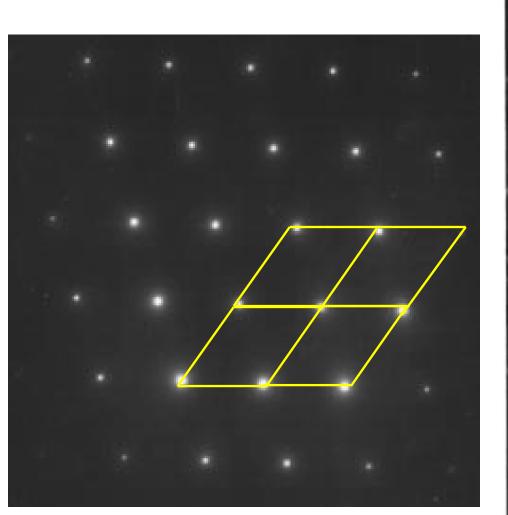
http://publcif.iucr.org/cifmoldb/gui/cifjmol.php?cifid=ck5037

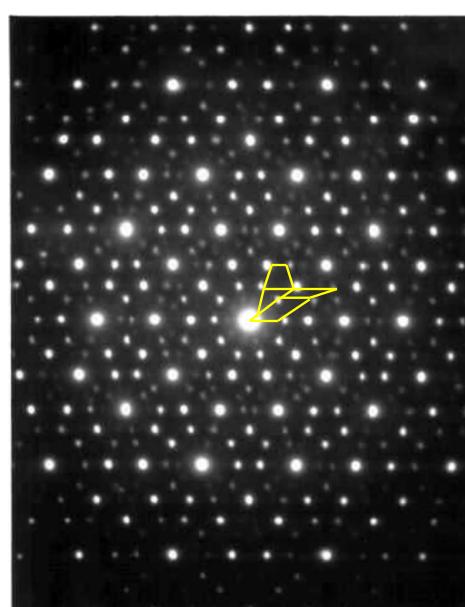
#### Quasicrystals

Angew. Chem. Int. Ed. 2011, 50, 10775 – 10778

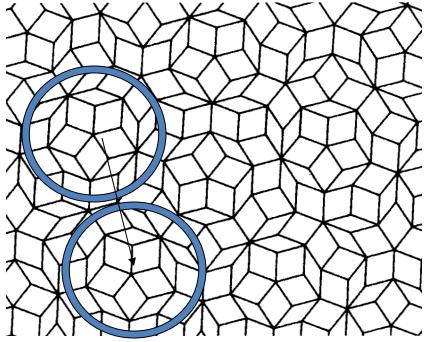


**Figure 1.** Electron diffraction patterns of the Al-Mn quasicrystal taken along different symmetry directions indicating icosahedral diffraction symmetry. Reprinted from Ref. [1].





## Penrose tiling: A 2-D incommensurate modulated structure (quasicrystal)



Roger Penrose
British Mathematician

Nobel Prize – 2011
Dan Shechtman
Technion - Israel
Institute of Technology,
Haifa, Israel

